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q=cache:feUWWVTetnQJ:en.wikipedia.org/wiki/Anthranilic_acid+anthranilic+acid+amino+acid&hl=en&ct=clnk&cd=13&gl=us

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These search terms have been highlighted: **anthranilic acid amino** These terms only appear in links pointing to this page: **acid**

Anthranilic acid

From Wikipedia, the free encyclopedia

		 Anthranilic	Acid
		0	_0

OH

			General
Systematic	. ,	• •	

name	2-ai	mın	ob	en	ZO	ic a	cid	
0.1	•.		•	1			• • •	

Other names	vitamin L1, Anthranilate
Molecular	CHNO

formula	$C_7H_7NO_2$			
Molor moss	127 126 a/mol			

CAS number [118-92-3]

Properties

Solubility in	0.6 a/I		
water	0.0 g/L		
0 1 1 111	T T		

Solubility in	Hot water
Density	1.4 g/cm ³

Melting point	146-148 °C
DOINL	

Boiling point Sublimes

Hazards

External (html)

MSDS

(http://newsearchus.chemexper.com/cheminfo/servlet/org.dbcreator.MainServlet?action=PowerSearch&query=msds. msdsID%

3D749&sort=&target=msds&from=0&realQuery=rn.value%3D%3D%22118-92-3%

22&searchTemplate=rn.value%3D%3D%3F&searchValue=118-92-

	3&history=off&options=brandqtyoffer&format=ccd)
EU classification	
R-phrases	R36 R37
S-phrases	S26 S39 .
Flash point	>150 °C
LD ₅₀ oral, mouse	1400 mg/kg
LD ₅₀ oral, rat	5410 mg/kg.
	Except where noted otherwise, data are given for materials in their standard state (at 25 °C, 100 kPa) Infobox disclaimer and references

Anthranilic acid has the formula $C_7H_7NO_2^{[1]}$ and is an odorless white to yellow crystalline acid, which is freely soluble in water. It sublimes at a temperature of 146-148°C.^[2]

Anthranilic acid is an aromatic amine which occurs physiologically as a metabolite of the amino acid tryptophan. It can also be made synthetically. [3]

Anthranilic acid is used as an intermediate for production of dyes, pigments and saccharin. It and its esters are used in preparing perfumes, pharmaceuticals and UV-absorber as well as corrosion inhibitors for metals and mold inhibitors in soya sauce.^[4] Anthranilic acid is also known unoffically as "vitamin L", a vitamin required for lactation in human females.^[3]

Anthranilic acid can be used in organic synthesis to generate the benzyne intermediate.

See also

Kynureninase

References

- ^ MSDS (http://newsearchus.chemexper.com/cheminfo/servlet/org.dbcreator.MainServlet? action=PowerSearch&query=msds._msdsID%
 3D749&sort=&target=msds&from=0&realQuery=rn.value%3D%3D%22118-92-3%
 22&searchTemplate=rn.value%3D%3D%3F&searchValue=118-92-3&history=off&options=brandqtyoffer&format=ccd)
- 2. ^ IPCS (http://www.inchem.org/documents/icsc/icsc/eics1295.htm)
- 3. ^ a b Vitamin L (http://www.greatvistachemicals.com/vitamins-vitamin/anthranilic-acid-vitamin-l.html)
- 4. ^ Benzoic Acid Derivatives (http://www.chemicalland21.com/specialtychem/finechem/o-AMINOBENZOIC%20ACID.htm)

External links

Links to external chemical sources

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Categories: Benzoic acids | Aromatic amines | Vitamins | DEA List I chemicals | Amine stubs

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